

Shedding Light on Energy Episode 2: Measuring Energy Name: _____

Part A

- In a light globe, _____ energy is transformed into _____ energy.
- In a wood fire, the _____ energy in the wood is transformed mainly into _____ energy and _____ energy.

Part B

- Fill in the table below.

Quantity	Unit Name	Unit Symbol	Quantity	Unit Name	Unit Symbol
length	metre	m	temperature		
energy			volume		L
mass		kg	time		s

- If 1 kilogram of water heats up from 20°C to 21°C it has absorbed _____ Joules of energy.
- If 1 kilogram of water heats up from 20°C to 22°C it has absorbed _____ Joules of energy.
- If 1 kilogram of water heats up from 40°C to 60°C it has absorbed _____ Joules of energy.

7. 1000 J = _____ kJ, 4200 J = _____ kJ, 7.9 kJ = _____ J, 0.5 kJ = _____ J.

- If 1 kilogram of water heats up from 50°C to 51°C it has absorbed _____ Joules of energy.
- If 2 kilograms of water heats up from 50°C to 51°C it has absorbed _____ Joules of energy.

- The equation to calculate the amount of energy absorbed by water is:

Energy absorbed = _____

- Fill in the table below.

Mass of water (kg)	Initial Temperature (°C)	Final Temperature (°C)	Temperature Change (°C)	Energy Absorbed (J)	Energy Absorbed (kJ)
1	40	60			
1	20	100			
2	20	100			
5	15		85		

Part C

- Our bodies get the energy that they need from _____.

13. Describe what energy intake is. _____

- Our bodies require about _____ Joules of energy per second when we are sitting down.

15. List 3 processes within our bodies or activities that need energy to occur. _____

- Fill in the table below.

ACTION	Approximate Energy Expenditure			
	Joules/second (J/s)	Joules/minute	kilojoules/minute	kilojoules per hour
Sitting still	100	6,000	6	360
Standing still	120			
Walking 3 km/hr	210			
Walking 5 km/hr	300			
Running 9 km/hr	700			
Running 16 km/hr	1300	78,000	78	4680
Cycling 9 km/hr	300			
Cycling 16 km/hr	500			

- Approximately how much energy do we typically expend per day? _____

18. How do scientists determine how much energy we use to perform different actions? _____

Part D

19. A typical person needs an energy intake of about _____ per day.
 20. Fill in the table below.

Nutrient	Energy Content	
	kilojoules/gram (kJ/g)	kilojoules/kilogram (kJ/kg)
Carbohydrates		
Fats and Oils		

21. A kilogram of carbohydrates contains enough energy to fuel a typical person for about _____ days.
 22. Fill in the table below. (The calculation is a two-step calculation.)

Food	Amount	Energy Content (total) (kilojoules, kJ)	Energy Content (kJ/100 mL or kJ/100 g)
milk	250 mL	650	
bread (2 slices)	76 grams	740	
egg (boiled or raw)	55 g	610	
avocado (1 whole)	220 g	1300	

23. How is the energy content in food determined? Use a diagram and text.

24. (a) A typical egg contains about _____ **kJ** of energy. Walking at 5 km/hr requires about _____ **kJ/min**.
 (b) How much walking time (in minutes) will it take to use up the chemical energy in the egg?

25. A 50 g block of chocolate contains 1125 **kJ**. How much time will you have to run (at 9 km/hr, which uses up about _____ **kJ/min**) to burn it off?

Part E

26. What is cellular respiration? _____

27. Complete the basic chemical equation that describes cellular respiration.



28. All of the _____ produced in cellular respiration and a lot of the _____ are expelled from the body via _____

29. If, over time, our energy intake is greater than our energy expenditure, we _____.
 30. If, over time, our energy intake is less than our energy expenditure, we _____.
 31. When people lose weight, where do the atoms that had made up the fat molecules actually go? (see Q27)

32. If, over time, our energy intake is the same as our energy expenditure, we _____.
 33. Explain why people often put on extra weight (fat) as they get older. _____

34. Explain how people (who are wishing to lose weight) can reduce their energy intake without eating less food.

